

AN INTRAMOLECULAR DIELS-ALDER REACTION.  
A SIMPLE SYNTHESIS OF  $\Upsilon$ -APOPICROPODOPHYLLIN.

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$\Upsilon$ -APOPICROPODOPHYLLIN (II) was isolated as a secondary degradation product resulting from dehydration of the naturally occurring, tumor-necrotizing compound podophyllotoxin.<sup>1</sup> II was found to be identical with a compound previously synthesized by Haworth and Richardson<sup>2</sup> by a multiple-step procedure, which was investigated further by Schrecker and Hartwell.<sup>3</sup> Interest in general syntheses of II and its analogous lignans of the 4-aryltetrahydronaphthalene type<sup>4</sup> has led us to investigate the possibility of effecting a simple intramolecular Diels-Alder condensation between two open-chain unsaturated Ar-C-C-C units (lignan building blocks), joined together at the  $\Upsilon$ -carbons in order to facilitate close proximity of these units. Needed to form II by such a scheme is

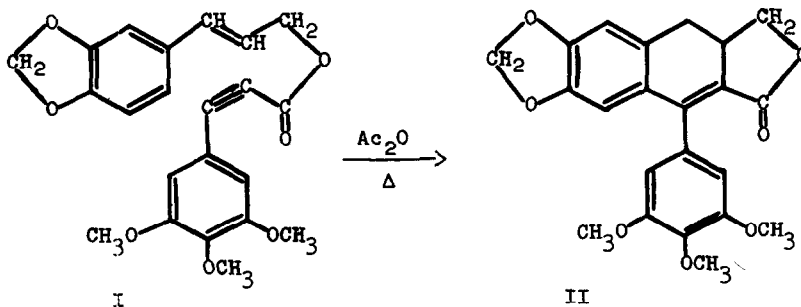
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<sup>1</sup> A.W. Schrecker and J.L. Hartwell, J. Amer. Chem. Soc. 74, 5676 (1952).

<sup>2</sup> R.D. Haworth and T. Richardson, J. Chem. Soc. 348 (1936).

<sup>3</sup> A.W. Schrecker and J.L. Hartwell, J. Amer. Chem. Soc. 74, 5672 (1952).

<sup>4</sup> W.M. Hearon and W.S. MacGregor, Chem. Rev. 55, 957 (1955).



I  
the ester 3,4-methylenedioxcinnamyl 3,4,5-trimethoxyphenylpropiolate (I).

Methyl 3,4,5-trimethoxycinnamate (III) was prepared by esterification of the corresponding acid (Aldrich Chemical Co.) with methanolic hydrogen chloride. The crude dibromo ester (m.p. 105-108°, 87% yield), resulting from treatment of a chloroform solution of III with bromine, was dehydrobrominated with alcoholic potassium hydroxide to form 3,4,5-trimethoxyphenylpropionic acid (IV), m.p. 141.5-142°,  $\nu_{\max}$  (nujol) at 1655 and 2140  $\text{cm}^{-1}$ , 37% yield from III (Found: C, 61.00; H, 5.26. Calc. for  $\text{C}_{12}\text{H}_{12}\text{O}_5$ : C, 61.01; H, 5.12). Treatment of IV with purified thionyl chloride at room temperature and then refluxing with equimolar amounts of pyridine and 3,4-methylenedioxcinnamyl alcohol (available from lithium aluminum hydride reduction of ethyl 3,4-methylenedioxcinnamate) gave I, m.p. 130-131°,  $\nu_{\max}$  (nujol) at 1690 and 2220  $\text{cm}^{-1}$ , 44% yield (Found: C, 66.78; H, 5.21. Calc. for  $\text{C}_{22}\text{H}_{20}\text{O}_7$ : C, 66.66; H, 5.09). Refluxing I with acetic anhydride for six hours gave a 48% yield of racemic II, m.p. 252-253°,  $\nu_{\max}$  ( $\text{CHCl}_3$ ) at 1750  $\text{cm}^{-1}$  (no band at 2220), identical in m.p., mixture m.p., infrared and ultraviolet absorption

spectra with a sample of bona fide  $\Upsilon$ -apopicropodophyllin kindly furnished to us from the stock of Schrecker and Hartwell. Cyclization to II was also effected in smaller yield by slow evaporative distillation of I at 240° and 0.3 mm.

In a similar fashion was prepared the enynic ester 3,4-methylenedioxcinnamyl 3,4-dimethoxyphenylpropiolate (V), m.p. 142-143°,  $\nu_{\max}$  (nujol) at 1710 and 2240  $\text{cm}^{-1}$ , 36% yield (Found: C, 68.71; H, 5.19. Calc. for  $\text{C}_{21}\text{H}_{18}\text{O}_6$ : C, 68.84; H, 4.95). This was cyclized with acetic anhydride to produce 1-(3,4-dimethoxyphenyl)-3-hydroxymethyl-6,7-methylenedioxy-3,4-dihydro-2-naphthoic acid lactone (VI), m.p. 222-223°,  $\nu_{\max}$  ( $\text{CHCl}_3$ ) at 1750  $\text{cm}^{-1}$  (no band at 2240), (Found: C, 68.39; H, 5.02). As for II, VI has an NMR absorption band at 7.1 $\tau$  (methylene protons at the 4-position) but none at 2.4 $\tau$  (expected for the  $\beta$ -proton in the system  $\text{ArCH}=\text{C}-\overset{\text{O}}{\underset{\text{O}}{\text{C}}}$ ,<sup>5</sup> which would result if Diels-Alder cyclization involved the  $\text{ArC}=\text{C}$ -group acting as a diene and the  $\text{C}=\text{C}$  group acting as a dienophile). The cinnamyl cinnamate (dienic) esters corresponding to I and V,<sup>6</sup> on the other hand, show NMR bands at 2.4 $\tau$  and none at 7.1 $\tau$ .

Work on these and analogous cyclizations is continuing in our laboratory.

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<sup>5</sup> M.F. Zinn, T.M. Harris, D.G. Hill, and C.R. Hauser, J. Amer. Chem. Soc. **85**, 71 (1963).

<sup>6</sup> Unpublished work from this laboratory.